Saponins from *Cussonia bancoensis* and Their Inhibitory Effects on Nitric **Oxide Production**

Leon A. Tapondjou,[†] David Lontsi,^{*,†} Beibam L. Sondengam,[‡] Fazarna Shaheen,[§] Mohammad I. Choudhary,[§] Atta-ur-Rahman,[§] Fanie R. van Heerden,^{\perp} Hee-Juhn Park,^{∇} and Kyung-Tae Lee^{\bigcirc}

Department of Chemistry, Faculty of Science, University of Dschang, PO Box 67 Dschang, Cameroon, Department of Organic Chemistry, Faculty of Science, University of Yaounde I, PO Box 812, Yaounde, Cameroon, International Centre for Chemical Sciences, HEJ Research Institute of Chemistry, University of Karachi, Karachi 75270, Pakistan, Department of Chemistry and Biochemistry, Rand Afrikaans University, PO Box 524, Auckland Park 2006, South Africa, Division of Applied Plant Sciences, Sangji University, Wonju 220-702, Korea, and College of Pharmacy, Kyung-Hee University, Seoul 130-701, Korea

Received August 30, 2002

Two new triterpenoid saponins (1 and 2) were isolated from the stem bark of *Cussonia bancoensis* together with the known stigmasterol, ursolic acid, 23-hydroxyursolic acid (3), and 3β -hydroxylup-20(29)-en-28oic acid. On the basis of their spectroscopic data and on chemical transformations, the structures of the new saponins have been established as $3-O(\alpha-L-arabinopyranosyl)-23$ -hydroxyursolic acid (1) and $3-O-\alpha$ $(\beta$ -D-glucopyranosyl)-23-hydroxyursolic acid (2). In a nitric oxide (NO)-production bioassay, compound 3 exhibited significant NO inhibitory activity, while compounds 1 and 2 were less potent than 3.

Cussonia bancoensis Aurev. & Pellegr. (Araliaceae) is a medium-sized tree of 15–25 m in height. It is found mostly in the dense humid forest and is widespread from Ivory Coast to Nigeria.¹ It is used in Nigerian folk medicine for the treatment of dizziness and infertility in women.¹ Previous work on this species has revealed the presence of flavonoids² and triterpene saponins.³ Phytochemical investigations on other Cussonia species have described the presence of saponins having oleanolic acid and hederagenin as aglycons and D-glucose, L-rhamnose, L-arabinose, and D-xylose as sugar moieties.^{4,5}

In a systematic search for new and/or bioactive metabolites from plant origin we have investigated the methanolic stem bark extract of C. bancoensis. We have isolated two new saponins, 3-O-(a-L-arabinopyranosyl)-23-hydroxyursolic acid (1) and $3-O-(\beta-D-glucopyranosyl)-23-hydroxyur$ solic acid (2), along with four known triterpenoids, namely, stigmasterol, ursolic acid, 23-hydroxyursolic acid (3), and 3β -hydroxylup-20(29)-en-28-oic acid. These compounds were tested for their inhibitory effects on a LPS-stimulated macrophage cell line (Raw 264.7 cells) on nitric oxide (NO) production. NO has diverse physiological roles and also contributes to immune defenses against viruses, bacteria, and other parasites. However, excess production of NO is associated with various diseases such as arthritis, diabetes, stroke, septic shock, autoimmune diseases, chronic inflammatory diseases, and atherosclerosis.⁶ Previous studies have revealed the potency of triterpenes and triterpenoidal saponins against cancer cell lines or the inhibition of NO production in LPS-activated macrophage cell lines.^{7–9}

The pulverized dried stem bark of *C. bancoensis* was extracted with methanol and concentrated under vacuum to dryness. The water-soluble portion of the methanol extract was successively extracted with ethyl acetate and *n*-butanol. The dried ethyl acetate-soluble fraction was then subjected to repeated column chromatography to afford four



known compounds. On the basis of the comparison of proton and carbon-13 nuclear magnetic resonance (1H and ¹³C NMR) spectra as well as direct comparison (co-TLC) with authentic samples, the known compounds were identified as stigmasterol, ursolic acid, 23-hydroxyursolic acid (3),^{10,11} and 3β -hydroxylup-20(29)-en-28-oic acid.¹²

Column chromatographic fractionation of the butanol extract gave four main fractions, and repeated column chromatography of the second fraction mainly afforded two saponins, **1** and **2**. The FABMS of **1** (negative-ion mode)

10.1021/np020412g CCC: \$25.00 © 2003 American Chemical Society and American Society of Pharmacognosy Published on Web 08/26/2003

^{*}To whom inquiries should be addressed. Tel: 237 7762780. E mail: dlontsi@uycdc.uninet.cm, dlontsi2000@yahoo.co.uk.

 [†] University of Dschang.
 [‡] University of Yaounde I.
 [§] University of Karachi.

¹ Rand Afrikaans University.

⁷ Sangji University.

[°] Kyung-Hee University.

Table 1. ¹³C NMR Spectral Data of Compounds 1-3 (at 100 MHz in C₅D₅N)

	1	2	3		1	2	3
carbon	$\delta_{\rm C}$ mult.	$\delta_{\rm C}$ mult.	$\delta_{\rm C}$ mult.	carbon	$\delta_{\rm C}$ mult.	$\delta_{\rm C}$ mult.	$\delta_{\rm C}$ mult.
1	38.8 t	38.7 t	39.0 t	21	36.8 t	37.4 t	37.0 t
2	28.6 t	28.4 t	26.9 t	22	31.0 t	31.0 t	33.0 t
3	81.8 d	82.1 d	74.2 d	23	64.8 d	64.5 d	64.6 t
4	43.4 s	42.8 s	43.8 s	24	13.6 q	13.3 q	12.7 q
5	48.5 d	48.6 d	48.0 d	25	16.2 q	15.2 q	15.6 q
6	18.1 t	18.3 t	13.3 t	26	17.4 q	17.2 q	17.5 q
7	33.1 t	33.0 t	33.0 t	27	23.8 q	23.6 q	24.2 q
8	39.9 s	39.8 s	40.0 s	28	179.8 s	179.6 s	179.7 s
9	47.4 d	48.5 d	48.5 d	29	17.3 q	17.1 q	17.1 q
10	37.4 s	37.3 s	37.1 s	30	21.3 q	21.0 q	21.1 q
11	24.8 t	23.7 t	23.9 t	sugar moieties	arabinose	glucose	•
12	125.6 d	125.3 d	125.6 d	e		0	
13	139.4 s	139.5 s	139.2 s	1′	106.4 d	105.9 d	
14	43.4 s	43.2 s	43.0 s	2'	73.8 d	75.9 d	
15	31.0 t	29.8 t	29.0 t	3′	74.6 d	78.7 d	
16	26.0 t	24.6 t	24.5 t	4'	69.5 d	71.8 d	
17	47.4 s	47.2 s	47.0 s	5′	66.8 d	78.4 d	
18	53.5 d	52.1 d	52.2 d	6′	62.9 d		
19	39.7 d	39.1 d	39.2 d				
20	39.4 d	39.2 d	39.0 d				

gave peaks at $m/z 603 [M - H]^-$ and 471 [(M - H) - 132]⁻, respectively, suggesting a molecular weight of 604 and the loss of a pentose moiety. Its IR spectra exhibited broad vibration bands at 3480 cm⁻¹ (hydroxy groups) and 1700 cm^{-1} (acid group). The EIMS of compound 1 showed ion peaks at *m*/*z* 472, 248, 203, 175, and 133, which indicated that the aglycon is an amyrin derivative having a free carboxylic group at C-17.^{13,14} Its ¹³C NMR spectrum (Table 1) showed signals at δ 125.6 (C-12) and 139.2 (C-13), characteristic for an urs-12-ene derivative,¹¹ which was further confirmed by its acid hydrolysis to give 23-hydroxyursolic acid (co-TLC with 3, ¹H NMR and EIMS) and L-arabinose (co-TLC with authentic sample). The sugar moiety was attached to the aglycon at position C-3 since the ¹³C NMR signal of the above carbon atom appeared at 81.8 ppm in compound 1, whereas the chemical shift of the same carbon atom appeared at 74.2 ppm in compound 3.^{10,11} The occurrence of a signal at 179.8 ppm showed the presence of a free carboxylic group at C-17. Further signals at 106.6, 73.8, 74.6, 69.5, and 66.8 assigned to C-1', C-2', C-3', C-4', and C-5' were in agreement with an arabinopyranosyl moiety.15,16

In the ¹H NMR spectrum of **1** the anomeric proton signal appeared at 4.95 ppm as a doublet with the coupling constant of 7.1 Hz, indicating its α -configuration.¹⁷ Acetylation of **1** in the usual manner provided the tetraacetate **4** with a downfield shift of the AB system signals of the two protons corresponding to CH₂OH-23, thus confirming the location of the arabinosyl moiety at C-3 rather than at C-23. Therefore, the structure of compound **1**, which is described here for the first time, was established as 3-*O*-(α -L-arabinopyranosyl)-23-hydroxyursolic acid.

Compound **2** was obtained as a white amorphous powder, and its molecular weight was deduced from its negative HRFABMS as 634. The main peaks observed in the FABMS (negative-ion mode) were at m/z 633 $[M - H]^-$ and 471 $[(M - H) - 162]^-$, with the latter corresponding to the loss of a hexose moiety. Its EIMS and IR spectra were very close to those of **1**, suggesting that these two compounds have the same aglycon. This was further confirmed on acid hydrolysis, which also afforded 23-hydroxyursolic acid (co-TLC) and D-glucose (co-TLC with authentic sample), instead of arabinose, as was the case with **1**. The ¹³C NMR spectrum of compound **2** (Table 1) was similar to that of **1** except for the presence of one carbon atom more than that of **1** (δ 60–90 region). The anomeric carbon signal appeared
 Table 2. Effect of Compounds Isolated from Cussonia bancoensis on NO Production by LPS-Induced Raw 264.7 Macrophages^a

	IC ₅₀ (μΜ)				
compound	NO inhibition	cytotoxicity for macrophages			
1	47.9	124			
2	22.1	126			
3	2.6	11			
ursolic acid	46.9	58.5			
L-NIL ^{b}	18.6	> 200			

 a The culture media were collected for nitrite determination. Results are the mean from four independent experiments. IC₅₀ is defined as the concentration that resulted in 50% decrease in NO production. b L-NIL (L- $N_6\text{--}1\text{-iminoethyl})$ lysine was used as a standard compound.

at 105.8 ppm and its proton at 4.94 ppm as a doublet with the coupling constant of 7.2 Hz, indicating its β -configuration.¹⁸ The acetylation of **2** afforded the pentaacetate **5**. The ¹H and ¹³C NMR data of the six methyl signals (Me-24–27, -29, and -30) of **1** were in agreement with those of the literature values of **3** (the aglycon of **1**) and related compounds.¹⁰ In light of the above findings, the structure of **2** was elucidated as the new compound 3-*O*-(β -Dglucopyranosyl)-23-hydroxyursolic acid.

In the nitric oxide production inhibition experiment (Table 2), 23-hydroxyursolic acid (3) exhibited the highest NO inhibitory activity (IC₅₀ 2.6 μ M) against LPS-induced macrophage cells. It was even more potent than the standard compound (L-NIL, IC₅₀ 18.6 μ M). Compounds 1 and 2 also showed activity, although they were less potent than 3. These findings are in agreement with the work of Ryu et al.,¹⁹ who recently reported the in vitro anti-inflammatory effect of other hydroxyursolic acids.

Experimental Section

General Experimental Procedures. Melting points were determined on a Yanaco micro-melting point apparatus. Optical rotations were measured on a JASCO DIP-360 digital polarimeter and IR spectra recorded on a JASCO A-302 spectrophotometer. The ¹H NMR spectra (δ ppm, *J* in Hz) were recorded in C₅D₅N or CD₃OD on a Bruker AM-400 spectrometer (400 MHz), while ¹³C NMR spectra were recorded in the same solvent on an AM-400 instrument at 100 MHz with tetramethylsilane (TMS) as internal standard. EIMS and FABMS were taken on MAT 311A and JMS HX-110 mass spectrometers, respectively. The purity of the samples was

checked on TLC (Si gel, precoated plates, Merck, $PF_{254},$ 20 \times 20 cm, 0.25 mm). Solvents were distilled prior to use.

Plant Material. The stem bark and leaves of *C. bancoensis* were collected in Bafou village (Menoua Division, Western Province of Cameroon) in April 2000. The plant material was identified by Dr. Zapfack Louis of the Department of Plant Biology of the University of Yaounde I. Specimens documenting the collection are deposited at the Cameroon National Herbarium (ref 16896/SRF/CAM).

Extraction and Isolation. The dried and pulverized stem bark of *C. bancoensis* (4 kg) was extracted with MeOH at room temperature for 3 days, and the MeOH extract concentrated to dryness under reduced pressure. The residue obtained (320 g) was suspended in water and extracted with ethyl acetate and *n*-butanol to afford, after evaporation, 120 and 135 g of an EtOAc extract and a *n*-BuOH extract, respectively.

A part of the EtOAc-soluble extract (75 g) was submitted to silica gel column chromatography using hexane-EtOAc mixtures of increasing polarity for elution. A fraction that was eluted with hexane-EtOAc (9:1) yielded stigmasterol (160 mg). Ursolic acid (210 mg) was obtained directly in pure form as white needles [mp 278–280 °C (lit. 280–282 °C, $[\alpha]_D$ +72° (CHCl₃)²⁰] from the fraction eluted with hexane-EtOAc (8.5: 1.5). 23-Hydroxyursolic acid (3) and 3β -hydroxylup-20(29)-en-28-oic acid were poorly separated and were obtained as a mixture from the fraction eluted with hexane-EtOAc (3:2). Separation of both substances was achieved by further purification in a silica gel column continuously eluted with the mixture EtOAc-hexane (7:3): Compound 3 [mp 280-282 °C (hexane–EtOAc), (lit. 283–287 °C, $[\alpha]_D^{25}$ +64° (c 0.27, MeOH)¹⁰] and 3*β*-hydroxylup-20(29)-en-28-oic acid [mp 270 °C (hexane-EtOAc), lit. 263–267 °C (MeOH/CHCl₃). $[\alpha]^{21}_{D}$ +88° (*c* 0.075, $C_5D_5N^{12}$ were then obtained in pure form as white needles, respectively, in the following amounts (7 g and 600 mg).

A part of the butanol-soluble extract (50 g) was subjected to silica gel column chromatography and eluted with the mixtures of CH_2Cl_2 –MeOH of increasing polarity. Four main fractions were obtained. The first one afforded, after repeated silica gel column chromatography, eluted with mixtures of CH_2 - Cl_2 –MeOH (4:1 to 3:2), compounds **1** (300 mg) and **2** (200 mg).

3-*O* (α-L-Arabinopyranosyl)-23-hydroxyursolic acid (1): white amorphous powder from MeOH/CH₂Cl₂; $[α]^{31}_D + 65.2^{\circ}$ (*c* 0.046 g/100 mL, MeOH); IR (KBr) ν_{max} 3480, 2990, 1700, 1640, 1450, 1051 cm⁻¹; ¹H NMR (C₅D₅N, 400 MHz) δ 0.92 (3H, d, *J* = 6.7 Hz, H-29), 0.94 (3H, d, *J* = 6.7 Hz, H-30), 0.95 (3H, s, H-24), 0.97 (3H, s, H-25), 1.00 (3×H, s, H-26), 1.20 (3H, s, H-27). 2.60 (IH, d, *J* = 11.2 Hz, H-18), 3.68 (1H, d, *J* = 12.1 Hz, H-23A), 3.69 (1H, dd, *J* = 11.0, 1.6 Hz, H-5A'), 4.00 (1H, dd, *J* = 8.9, 3.4 Hz, H-3), 4.24 (1H, dd, *J* = 11.0, 3.0 Hz, H-5B'), 4.26 (1H, m, H-4'), 4.26 (1H, d, *J* = 12.1 Hz, H-23B), 4.46 (1H, dd, *J* = 9.0, 7.2 Hz, H-2'), 4.95 (1H, *J* = 7.2 Hz, H-1'). 5.45 (br t, H-12); ¹³C NMR, see Table 1; EIMS *m*/*z* 472 (C₃₀H₄₈O4, aglycon), 248, 203, 175; FABMS (negative) *m*/*z* 603 [M - H]⁻, 471 [(M - H) - 132]⁻; HRFABMS (negative) *m*/*z* 603.8112 [C₃₅H₅₆O₈ - H]⁻, calcd 603.8143.

3-*O*-(β-D-Glucopyranosyl)-23-hydroxyursolic acid (2): white amorphous powder from MeOH/CH₂Cl₂; [α]_D³¹ +43.4° (*c* 0.046 g/100 mL, MeOH); IR (KBr) ν_{max} 3490, 2990, 1695, 1635, 1450, 1080, 1050 cm⁻¹; ¹H NMR (C₅D₅N, 400 MHz) δ 0.91 (6H, d, *J* = 6.7 Hz, H-29 and H-30), 0.95 (3H, s, H-24), 0.96 (3H, s, H-25), 1.04 (3H, s, H-26), 1.16 (3H, s, H-27), 2.62 (1H, d, *J* = 11.2 Hz, H-18), 3.73 (1H, d, *J* = 12.3 Hz, H-23A), 3.95 (1H, m, H-5'), 4.05 (1H, dd, *J* = 8.8, 3.4 Hz, H-3), 4.22 (1H, d, *J* = 12.3 Hz, H-23B), 4.22 (1H, dd, *J* = 5.6, 11.4 Hz, H-6A'), 4.23 (1H, dd, *J* = 8.5, 8.7 Hz, H-4'), 4.25 (1H, dd, *J* = 8.5, 8.7 Hz, H-3), 4.35 (1H, dd, *J* = 7.2, 8.5 Hz, H-2'). 4.45 (1H, dd, *J* = 2.2, 11.4 Hz, H-6B'), 4.98 (1H, d, *J* = 7.2 Hz, H-1'), 5.41 (1H, br t, H-12); ¹³C NMR, see Table 1; FABMS (negative *m*/*z* 633 [M - H]⁻, 471 [(M - H) - 162]⁻; HRFABMS (negative) *m*/*z* 633.8251 [(C₃₆H₅₈O₉) - H]⁻, calcd 633.8218.

Acid Hydrolysis of 1 and 2. Saponins 1 and 2 (15 mg each) were refluxed with 4% H₂SO₄ (5 mL) in MeOH for 1 h. Each reaction mixture was then concentrated under reduced pressure to remove MeOH, diluted with H₂O, and filtered. The

precipitate of each was purified by recrystallization from MeOH to afford the same aglycon, 23-hydroxyursolic acid, as needles (mp 280–281 °C), which was identified by direct comparison with an authentic sample. Each filtrate was adjusted to pH 7 with BaCO₃ and filtered. The two filtrates were concentrated and examined by TLC (MeOH–CH₂Cl₂–AcOH–H₂O, 40:55:3:2). Arabinose from saponin **1** and glucose from saponin **2** were identified by co-TLC comparison with

authentic samples. 3-O-(α-L-Arabinopyranosyl)-23-hydroxyursolic Acid 23,2',3',4'-Tetraacetate (4). A solution of 1 (20 mg) in Ac₂Opyridine (1:1) was stirred overnight. The reaction mixture was evaporated to dryness and chromatographed on silica gel using hexane-EtOAc (1:1) to give a tetraacetate (4) as a white powder in hexane-EtOAc: IR (KBr) v_{max} 2990, 1740, 1690, 1450, 1380 cm⁻¹; ¹H NMR (C₅D₅N, 400 MHz) δ 0.80 (3H, s, H-24), 0.82 (3H, s, H-25), 0.94 (3H, d, J = 6.5 Hz, H-29), 0.99 (3H, d, J = 6.5 Hz, H-30), 1.00 (3H, s, H-26), 1.18 (3H, s, H-27), 1.95, 2.01, 2.15, and 2.22 (3H each, s, 4 × OAc), 2.62 (1H, d, J = 11.3 Hz, H-18H), 3.81 (1H, dd, J = 11.8, 4.7 Hz, H-3), 3.91 (1H, dd, J = 13.1, 1.5 Hz, H-5'ax), 4.01 (1H, d, J = 11.3 Hz, H-23A), 4.12 (1H, br d, J = 13.2 Hz, H-5'eq), 4.22 (1H, d, J =11.2 Hz, H-23B), 4.91 (1H, d, J = 7.5 Hz, H-1'), 5.40 (1H, br t, H-12), 5.52, 5.60, 5.70 (1H, m, H-23, H-2', H-4').

3-*O*-(β -D-Glucopyranosyl)-23-hydroxyursolic Acid 23,2', 3',4',6'-Pentaacetate. Compound 2 was acetylated and purified as described above to yield a pentaacetate (5) as a white powder (hexane/EtOAc): IR (KBr) ν_{max} 2995, 1745, 1690, 1450, 1383 cm⁻¹; ¹H NMR (C_5D_5N , 400 MHz) δ 0.78 (3H, s, H-24), 0.80 (3H, s, H-25), 0.93 (3H, d, J = 6.2 Hz, H-29), 0.99 (3H, d, J = 6.0 Hz, H-30), 1.02 (3H, s, H-26), 1.20 (3H, s, H-27), 1.98, 1.99, 2.05, 2.12, 2.22 (3H each, s, 5 × OAc), 2.65 (IH, d, J =11.1 Hz, H-18), 3.96 (1H, dd, J = 12.0, 4.8 Hz, H-3), 3.98 (1H, dd, J = 12.4 Hz, H-23A), 4.32 (IH, d, J = 12.4 Hz, H-23B), 4.08 (1H, dd, J = 12.2, 2.5 Hz, H-6'A), 4.25 (1H, dd, J = 12.2, 4.7 Hz, H-6'B), 5.18, 5.05, 4.93, 3.65 (1H, m, H-2', H-3', H-4', H-5'), 5.38 (1H, br t, H-12).

Bioassays. Cell Culture and Sample Treatment. The Raw 264.7 murine macrophage cell line was obtained from the Korea Cell Line Bank (Seoul, Korea). DMEM medium, fetal bovine serum (FBS), penicillin, and streptomycin were obtained from Life Technologies, Inc. (Grand Island, NY). L- N^{6} -(1-Iminoethyl)lysine (L-NIL) and *E. coli* lipopolysaccharide (LPS) were purchased from Sigma Chemical Co. (St Louis, MO). These cells were grown at 37 °C in DMEM medium supplemented with 10% FBS, penicillin (100 units/mL), and streptomycin sulfate (100 μ g/mL) in a humidified atmosphere of 5% CO₂. Cells were incubated with LPS at 1 g/mL for 24 h.

Nitrite Assay. Nitrite accumulation, an indicator of NO synthesis, was measured in the culture medium by the Griess reaction.⁹ Briefly, 100 μ L of cell culture medium was mixed with 100 μ L of Griess reagent [equal volumes of 1% (w/v) sulfanilamide in 5% (v/v) phosphoric acid and 0.1% (w/v) naphthylethylenediamine HCl] and incubated at room temperature for 10 min, and then the absorbance at 550 nm was measured in a microplate reader. Fresh culture medium was used as the blank in all experiments. The amount of nitrite in the samples was calculated from a sodium nitrite standard curve freshly prepared in culture medium.

Acknowledgment. This research project was supported by the Third World Academy of Sciences (Trieste, Italy), which is gratefully acknowledged. Dr. Zapfack Louis, Department of Plant Biology of the University of Yaounde I, is also acknowledged for his contribution to the identification of the plant material.

References and Notes

(1) Adjanohoun, E.; Ahiyi, M. R. A.; Aké Assi, L.; Dramane, K.; Elewude, J. A.; Fadoju, S. O.; Gbile, Z. O.; Goudoté, E.; Johnson, C. L. A.; Keita, A.; Morakinyo, O.; Ojewole, J. A. O.; Olatunji, A. O.; Sofowora, E. A. *Traditional Medicine and Pharmacopoeia, Contribution to Ethnobotanical and Floristic Studies in Western Nigeria*; Commission of Scientific and Technical Research/Organization of African Unity: Lagos, 1991.

- (2) Haruna, A. K.; Ilyas, M.; Mustapha, A. Ghana J. Chem. 1990, 1, 160-165
- (3) Haruna, A. K.; Ilyas, M.; Mustapha, A. Ghana J. Chem. 1994, 1, 401-403

- 405.
 (4) Dubois, M. A.; Ilyas, M.; Wagner, H. *Planta Med.* 1986, *52*, 80-83.
 (5) Chazan, J. B. *Phytochemisty* 1971, *10*, 2111-2116.
 (6) Bredt, D. S.; Snyder, S. H. *Annu. Rev. Biochem.* 1994, *63*, 175-195.
 (7) Lee, K. T.; Sohn, I. C.; Park, H. J.; Kim, D. W.; Jung, G. O.; Park K. Weight and the approximation of the second second
- (7) Lee, K. 1., Sonn, I. C., Park, H. S., Kin, D. W., Sung, G. O., Park K. Y. *Planta Med.* **2000**, *66*, 2111–2116.
 (8) Park, H. J.; Kwon, S. H.; Lee, J. H.; Lee, K. H.; Miyamoto, K.; Lee, K. T. *Planta Med.* **2001**, *67*, 118–121.
 (9) Kim, Y. K.; Park, S. J.; Ha, J. H.; Choi, J. W.; Park, H. J.; Lee K. T. *Biol. Pharm. Bull.* **2003**, *26*, in press.
- Shin, Dam, Dan, Doo, Dr. Phytochemistry 1989, 28, 644–647.
 Mahato, B. S.; Kundu, P. A. Phytochemistry 1994, 37, 1517–1575.
 Ikuta, A.; Itokawa, H. Phytochemistry 1988, 27, 2813–2815.

- (13) Djerassi, C.; Budzikiewicz, H.; Wilson J. M. Tetrahedron Lett. 1962, 7, 263–270.
- (14) Budzikiewicz, H.; Djerassi, C.; Williams, D. H. Structure Elucidation of Natural Products by Mass Spectrometry; Holden-Day: San Fran-
- cisco, 1964; Vol. 2, p 122.
 (15) Lavaud, C.; Voutquenne, L.; Massiot, G.; Men-Olivier, L. L.; Das, B. C.; Laprevote, O.; Serani, L.; Delaude, C.; Becchi, M. *Phytochemistry* **1998**, *47*, 441–449.
- (16) Pancharoen, O.; Tuntiwachwuttikul, P.; Taylor, W. C.; Picker, K. *Phytochemistry* **1994**, *35*, 987–992.
 (17) Ryu, S. Y.; Oak, M. H.; Yoon, S. K.; Cho, D. I.; Yoo, G. S.; Kim, T. S.; Kim, K. M. *Planta Med.* **2000**, *66*, 358–360.
 (18) Boiteau, P.; Pasich, B.; Rakoto, A. R. Les Triterpénoides en Physiologie Charles and C
- Vegetale et Animale, Gauthier Villar: Paris, 1964; p 184.

NP020412G